Mass Spec

| $\mathbf{m} / \mathbf{z}$ ratio | group |
| :--- | :--- |
| 88 | $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{O}$ is the empirical formula <br> and has a mass of 44 so <br> compound must have formula <br> $\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{O}_{2}$ |

IR:

| Wavenumber $\left(\mathbf{c m}^{-1}\right)$ | Characteristic group |
| :--- | :--- |
| 2980 broad | -OH |
| 1710 sharp | $\mathrm{C}=\mathrm{O}$ |

## NMR:

There are 4 carbons in the molecule.

